



NATIONAL RESEARCH UNIVERSITY

St. Petersburg State Polytechnical University

Founded in 1899





HISTORY











Founded by



Sergey Yu. Vitte - the Prime Minister of Russia (1903-1906), initiated the idea of setting up Polytechnic institutes in St.Petersburg, Warsaw, Kiev



Andrey G. Gagarin - the first Director of the Institute



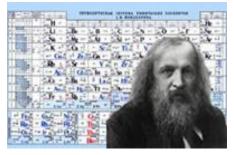


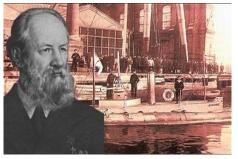


Founders

- **Dmitry I. Mendeleev** Chemistry
- Alexander S. Popov Radio Engineering
- Alexey N. Krylov Shipbuilding











HISTORY

Famous Scientific Developments

- Submarine "Dolphin", **1904**
- Plane "Po-2", **1928**
- Implementation of State Electrification program, **1926-1935**
- **u** Tank T-34, **1939**
- Nuclear Ice-breaker "Lenin", 1954
- Automated control systems on spaceships "Vostok", 1961
- Plane "Antey", **1965**





Famous Scientific Developments

- Space developments, **1966-1970**
- Radio telescope RATAN-600, **1968-1974**
- Construction of biggest Hydroelectric and Nuclear power stations, **1977**
- Launching thermonuclear synthesis device "TOCAMAC-16", **1988**
- Spacecrafts of RSC "Energia", **2000-s**
- Engines for "MIG" aircraft, helicopter "Black Shark", 2000-s
- Bioenergetics and hydrogen energy, 2000-s







Nobel Prize Winners

- Nicolay N.Semenov (Chemistry, 1956)
- Pyotr L. Kapitsa (Physics, **1978**)
- Zhores I. Alferov (Physics, 2000), Dean of the Faculty of Physics and Technology







St. Petersburg State POLYTECHNICAL UNIVERSITY



www.spbstu-eng.ru

UNIVERSITY TODAY





General Information

- Total area 102 hectares
- Total number of buildings **112** (total area of **386 061** sq.m)
- Faculties 21







In 2010 SPbSPU

тетербу государственный политехнический национальный исследовательский

was awarded the status of

"National Research University"

In 2010, **29** Russian Universities were awarded the status **"National Research University"** by the order of Russian Federation Government

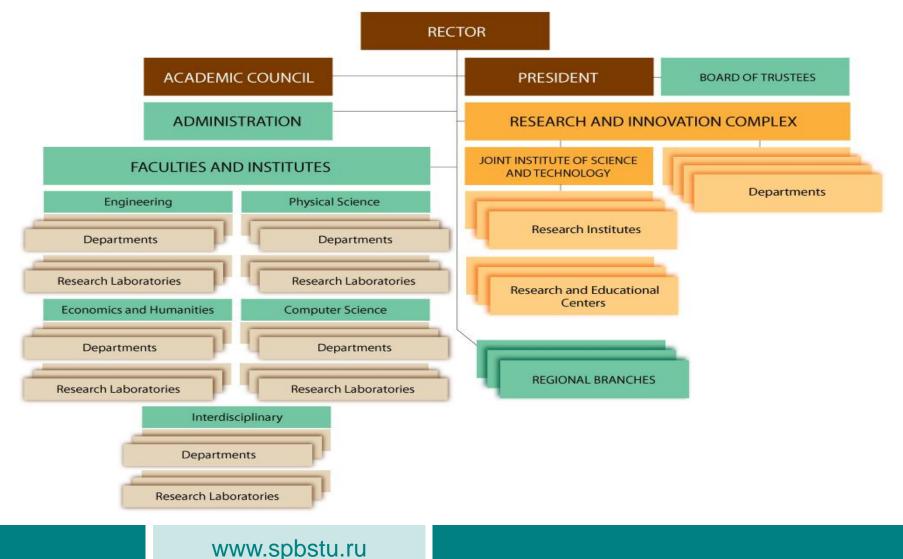




St. Petersburg State POLYTECHNICAL UNIVERSITY

UNIVERSITY TODAY

University Structure





Structural Units

- Joint Science and Technology Institute including
 - 8 Innovative Research Institutions
- 21 faculties and institutes
- Over 150 departments and 120 R&E laboratories
- 26 research and educational centers
- More than 20 small innovative enterprises
- Science park and incubating
- Faculty of foundation programs and institute of professional trainings
- 3 branches and 6 representatives
- St. Petersburg College of Information and Management

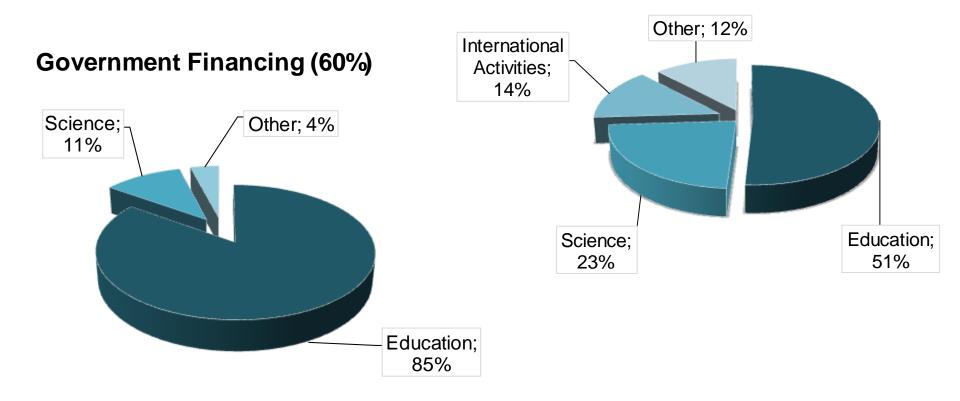




UNIVERSITY TODAY

Budget

Non-Government Financing (40%)





International Research Projects and Programs

- EU Framework Programs
- ENPI CBC
- International Science and Technology Center (ISTC)
- NATO-Russia Science for Peace Project
- Civilian Research and Development Foundation (CRDF)
- Forskningradet Cooperation Program with Russia (RUSSIA)
- National Science Foundation (NSF)
- National Fund of Canada





Cooperation with Foreign Companies







Joint Science and Technology Institute

According to the Program in 2010 was founded **Joint Science and Technology Institute** includes the following **Innovative Research Institutions**:

- Research Institute of Materials and Technologies
- Research Institute of Energy, Resources Saving and Environmental Technologies
- Research Institute of Nano-biotechnologies
- Research Institute of Electronic Systems
- Research Institute of Mathematical Modelling and Intelligent Control Systems
- Research Institute of Machine-Building Technologies (MashTeh)
- Research Institute of Computer Cybernetics
- Research Institute of Nuclear Physics





Nanotechnology Competence and Innovation

Research Institute of Materials and Technologies :

- Gas Phase Synthesis of Nanoparticles
- Li-ion Battery Technologies
- Refractory Metal Composites
- Polymer-Inorganic Nanocomposites
- Magnetic Nanocomposites
- LaB₆ contained heat resistive transparent films and coatings
- Solid Lubricants Nanoparticles





Technologies of nanopowders synthesis

- Gas Phase Synthesis of nanopowders (Fe, Fe-Co, W, WC, W(S,Se)2, Cu, Ni, TiO2, MgO, ZnO, carbon nanostructures, etc.)
- Spray-drying and thermal destruction of salt solutions (Cu-Al2O3, WC-Co-(VC, TaC))
- Electrochemical synthesis (LaB6, TiB2)
- Direct synthesis of carbon nanostructures on the surface of metallic powders (Fe, Cu, MgO, cement)
- Microwave synthesis (Al2O3, CuInSe2, spinels, etc)
- Mechanical Alloying (LaB6, HfC, ZrC, etc.)





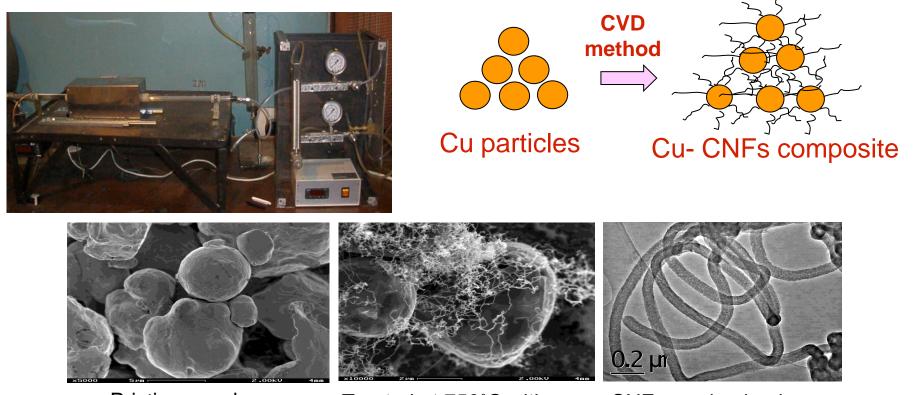
Potential Application Areas

- Magnetic recoding media
- Magnetic liquids
 - Medical and biological applications:
 - Magnetic Resonance Imaging
 - Magnetic Hyperthermia
 - Drag delivery systems
- Permanent magnets
 - Fillers for composite materials:
 - Magnetic plastics
 - ✓ Sensors
 - Low friction materials
- Nanoelectronics
- Pigments
- Catalysis





Copper – Carbon Composites



Pristine powder

Treated at 750°C with acetylene

CNFs synthesized at 750°C with acetylene



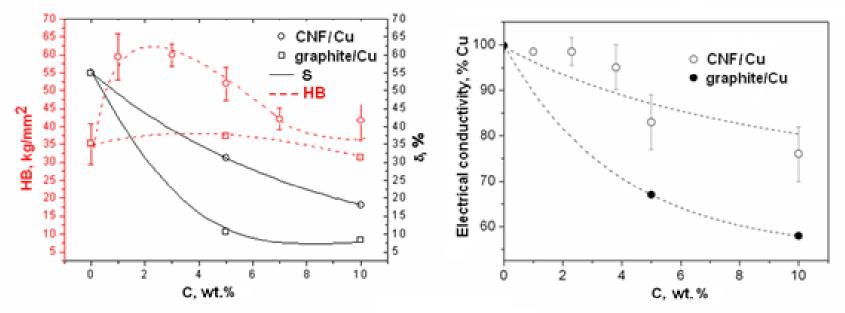


Sample Preparation and Properties

1, 3, 5, 7 and 10% -Cu/CNFs in pristine copper



Sintering: at 950°C, for 2 hours in H_2







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Mechanical and Electrical Properties



Antifriction materials:

rings, bushings, self lubricating bearings, flange gaskets.



Sliding electrical contacts:

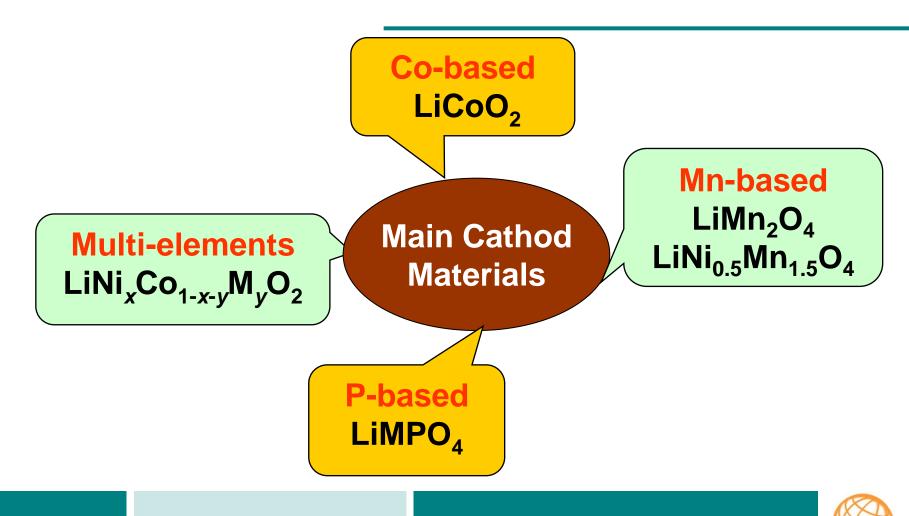
brushes of low power electrical machines, low-voltage generators



Lithium Polymer Battery (working imitation) Charging <u>e</u> 2 Discharging **e**⁻ Anode Cathode 4..... Li-NiCoMnO₂ Separator **Graphite or Carbon Polymer electrolyte** —Li; –Li⁺ (contains Li⁺)



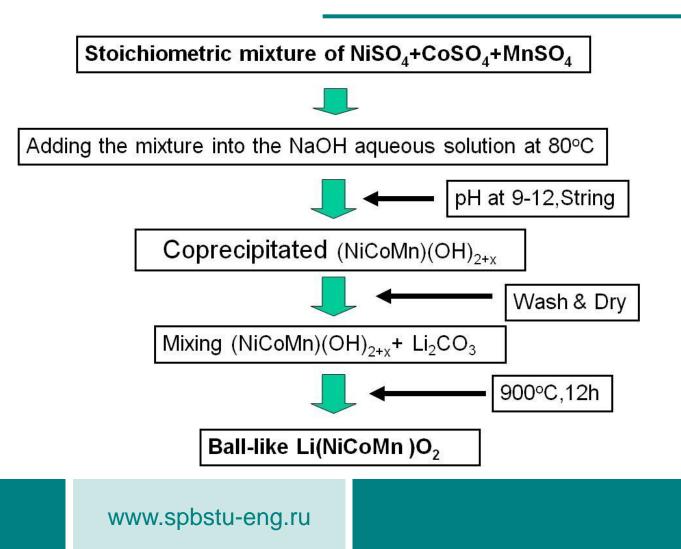
Cathode Materials





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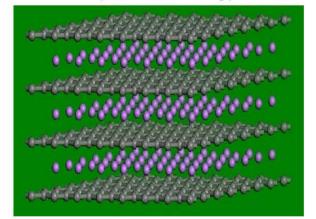
Cathode Material Preparation



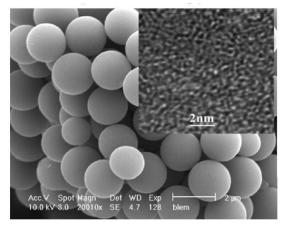




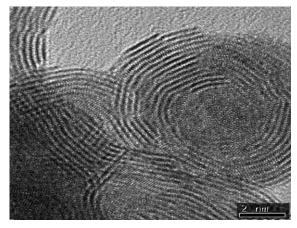
Anode Materials



Graphite (370 mAh/g)



Disordered carbon (200-500 mAh/g)



Layered halcogenide (>600 mAh/g)



Hybrid polymer-inorganic composite material

Outline

- Polyimide-based magnetic hard materials
- LaB₆ contained heat resistive transparent films and coatings
- Polycarbonate based materials for tribological applications

WHY Polymer-Inorganic NANOCOMPOSITES ?

- 1. Sufficient improvement of mechanical (modulus, strength) properties of polymers with small loading of nanoparticles
- 2. Sufficient improvement of thermal stability and flammability of polymer based materials
- 3. New functional properties (Magnetic, Electric, Transport, etc.)

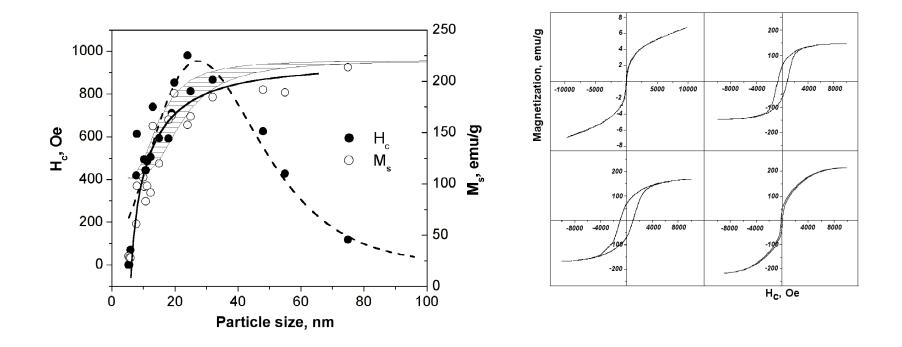
Nanoparticles used for polymers modifications

Nº	nanoparticles	Polymer matrix	Microstructure	Synthesis method
1	Iron-based magnetic nanoparticles	Polyimide	20 nm	Gas phase synthesis
2	LaB6 nanoparticles with the average size of 40-80 nm	Polyimide, polyurethan e		Thermal synthesis, electrochemical synthesis
3	WS ₂ , WSe ₂ nanoparticles	Polycarbon ate		Gas phase synthesis

Potential application areas

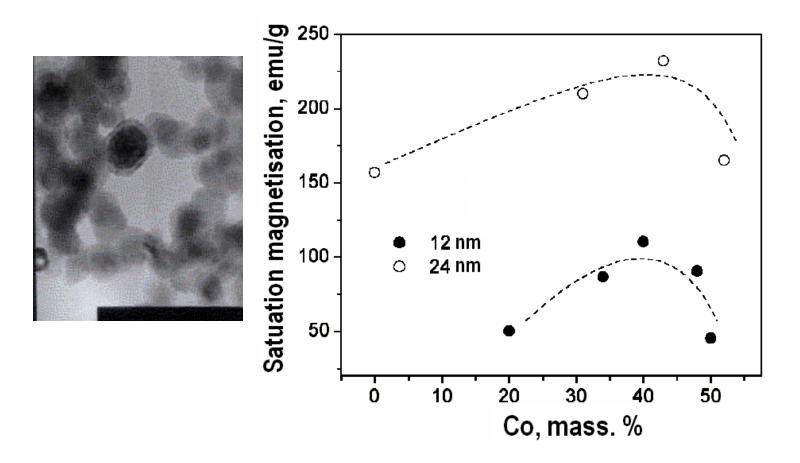
- Magnetic plastics;
- Magnetic shields for high frequency applications;
- Optical materials;
- Tribological applications;
- > Transparent heat-resistive materials.

Magnetic properties of iron based nanoparticles



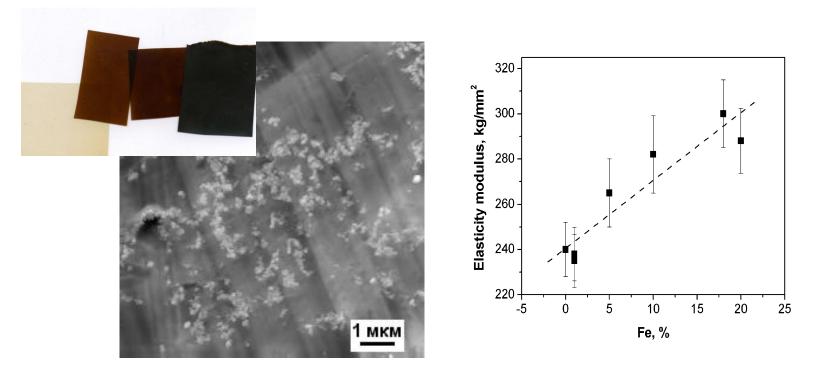
Coersivity (H_c) and magnetization (M_s) in the field of 10 kOe

Fe-Co alloyed nanoparticles



Coersivity of Ir-Co nanoparticles increased up to 1300 Oe;
The increasing of Co content up to 40% significantly increase saturation magnetization of nanoparticles (up to 245 emu/gr).

Polyimide-based composites

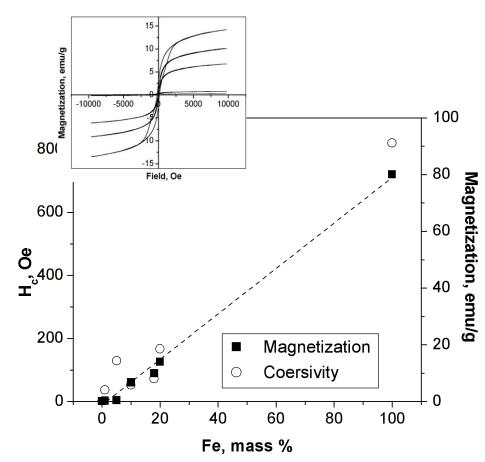


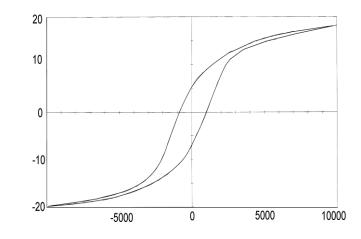
Advantages: 1) Uniform distribution of nanoparticles in amorphous matrix;

2) **Identical structure and chemical composition** of initial powder and powder in dispersed polymer;

3) High level of polymers' magnetic properties.

Magnetic properties



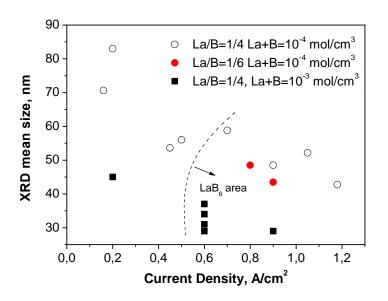


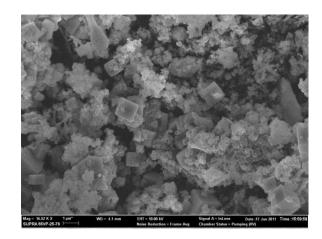
Nanoparticles of Fe-Co – polyimide composite solidified in magnetic field.

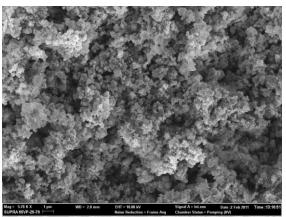
Coersivity is 890 Oe

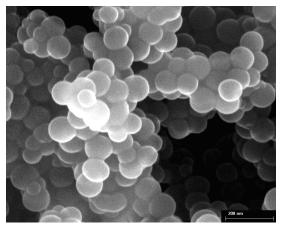
(Much higher as compare with cast permanent magnets)

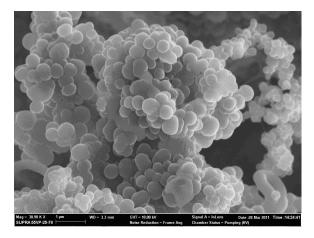
SEM micrographs and size of LaB6 nanopowders



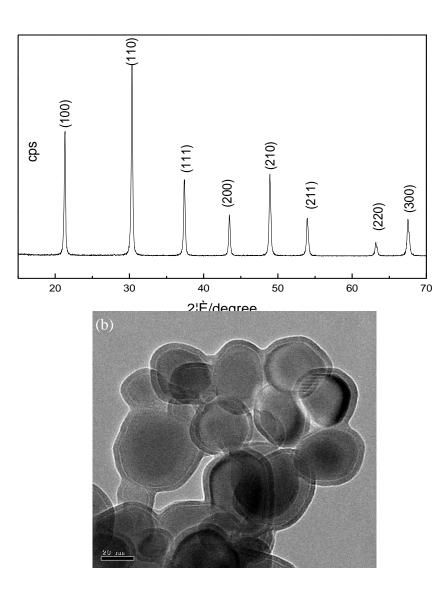


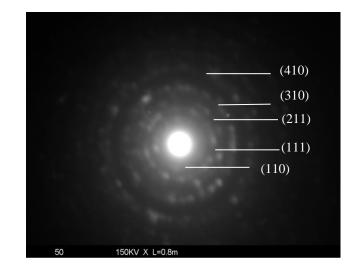


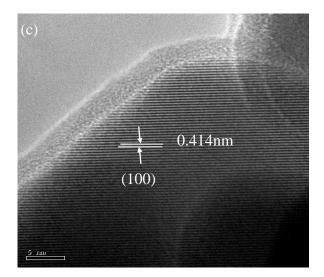




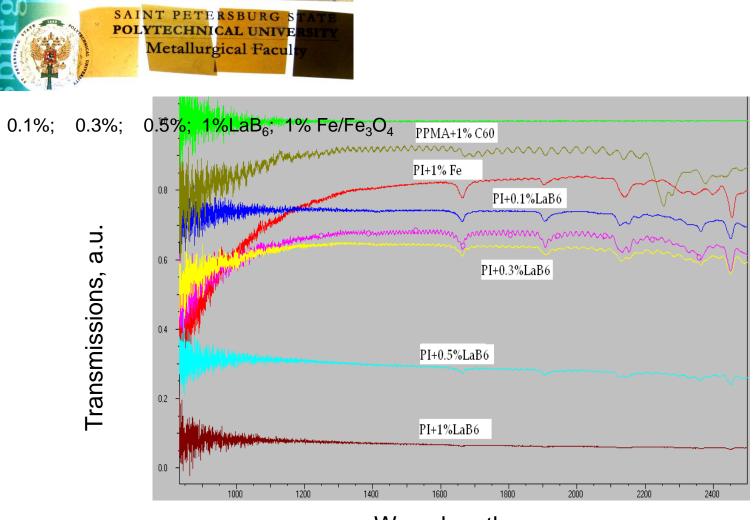
Structure of LaB₆. nanoparticles





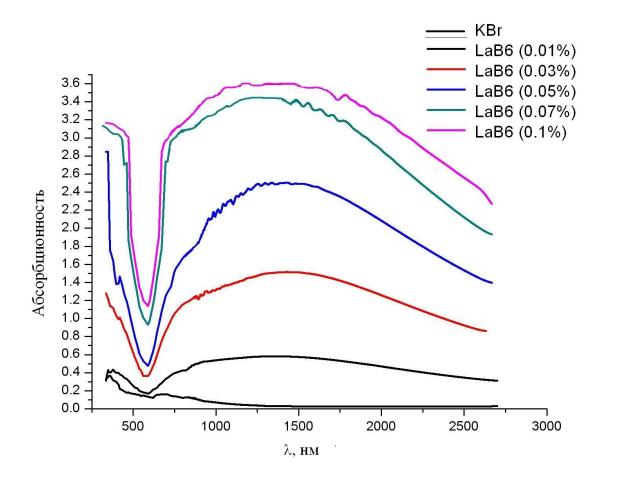


Comparison of IR spectra of polyimide films contained nanoparticles



Wave length, nm

Light absorption by LaB₆ in KBr matrix





PU based composites

Potential applications



Automobile glasses



Windows

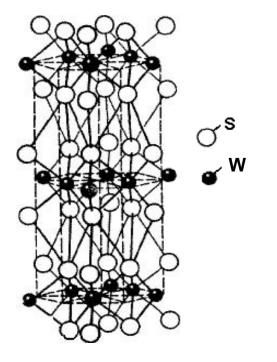


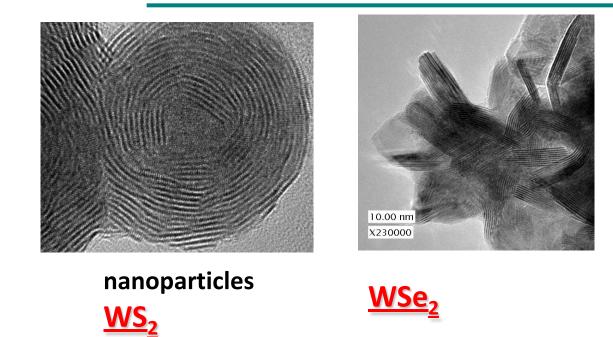


Wear and glasses for special applications



Solid Lubricant Nanoparticles





Low friction mechanism:

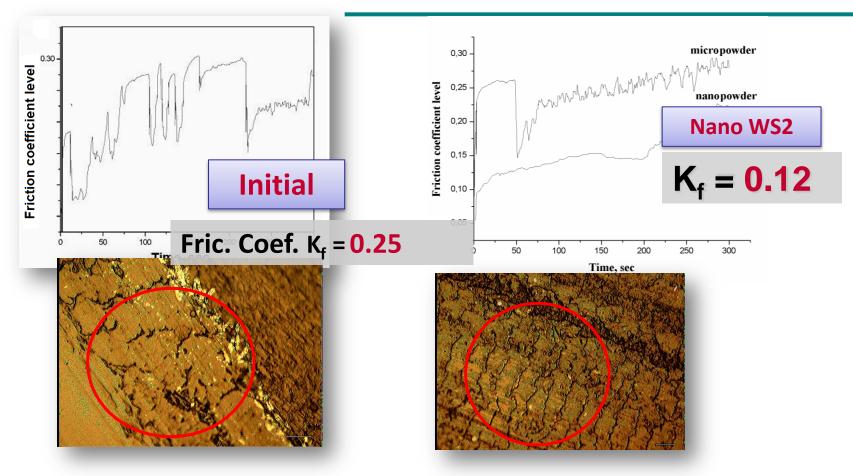
•Van der Waals forces between sulfur planes explain low inter-planar shear strength like so it works under high load and low speed.





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Polycarbonate-based nanocomposites









195251 Russia Saint-Petersburg 29, Polytechnicheskaya st. <u>www.spbstu-eng.ru</u> oleg@ftim.spbstu.ru





Thank you!



